

Crystal Distortion and Cation Distribution in the System $\text{La}_{1/3}\text{NbO}_3\text{--WO}_3$

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Synopsis. It has been found that WO_3 is soluble in $\text{La}_{1/3}\text{NbO}_3$ up to ca. 35 mol %. The solid solutions showed an ordered perovskite structure. The orthorhombic distortion changes into a tetragonal one at 20 mol %. The temperature dependence of the tetragonality of typical samples has been examined and the nature of the tetragonal distortion discussed.

$\text{La}_{1/3}\text{NbO}_3$ is an orthorhombically distorted structure containing two perovskite subcells.¹⁾ Iyer *et al.* reported that the perovskite-type unit cell contains a preferential distribution of La^{3+} ions on alternate c-planes, as shown in Fig. 1.²⁾ WO_3 , on the other hand, is well-known to have a perovskite structure without A-site ions. A-site deficient perovskites with a wide range of non-stoichiometry, such as $\text{La}_{1-x}\text{TiO}_3$ ³⁾ ($x=0.03\text{--}0.33$) and $\text{Sr}_{1-x}\text{NbO}_3$ ⁴⁾ ($x=0.05\text{--}0.30$), have been reported to be of simple perovskite structure type. $\text{La}_{2/3}\text{TiO}_3$ is an exception and has the same ordered structure as $\text{La}_{1/3}\text{NbO}_3$.⁵⁾ It is of interest crystallographically to understand how solid solutions between $\text{La}_{1/3}\text{NbO}_3$ and WO_3 can be formed and whether such ordered arrangement is maintained. The present paper describes the structural properties of this system.

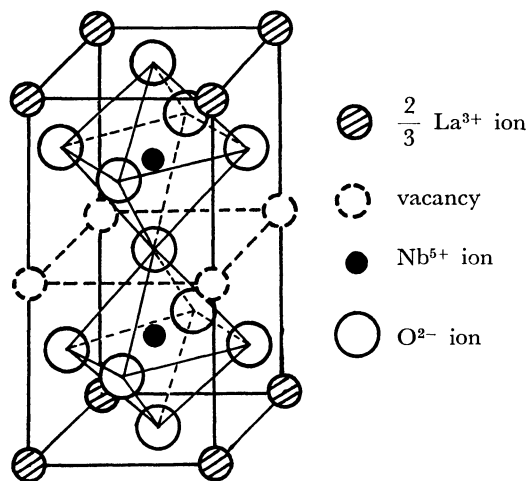
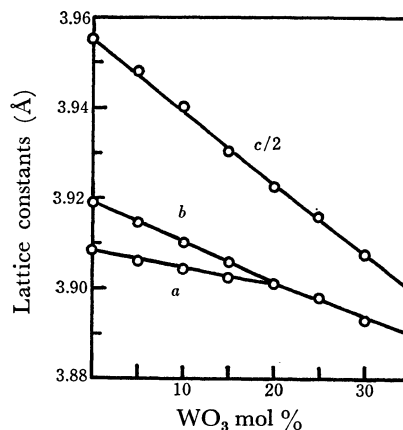
Experimental

The samples were prepared by conventional ceramic techniques, using La_2O_3 , Nb_2O_5 , and WO_3 as starting materials. Two-cycle firing was carried out in air at 1050—1270 °C for several hours, the samples being left to cool in the furnace. The products were analyzed by X-ray diffraction method, using $\text{Cu K}\alpha$ radiation. The lattice constants were determined from the best resolved lines of (004), (020), and (200), which appeared at $2\theta=46\text{--}47^\circ$. Observed intensities were obtained from the diffractometer tracing by weighing the relative areas under the peaks. The theoretical intensities

of the reflection lines were calculated according to a method previously reported.⁶⁾

Results and Discussion

The $\text{La}_{1/3}\text{NbO}_3\text{--WO}_3$ system showed single-phase patterns similar to that for $\text{La}_{1/3}\text{NbO}_3$ up to a composition containing 35 mol % WO_3 . In a 40 mol % WO_3 sample, a further phase appeared in small amounts besides the perovskite-like compound. The room temperature lattice constants are shown in Fig. 2. The crystal lattice became contracted by the partial substitution of W^{6+} ions for Nb^{5+} ions. The $c/2a$ ratio decreased with increasing amounts of WO_3 . The orthorhombic distortion changed into a tetragonal one at 20 mol % WO_3 and the tetragonality decreased with further increase of WO_3 , but the perovskite subcell did not become cubic. The decrease in crystal distortion is probably attributable to the size effect of B-site ions. The intensities of superstructure lines were also observed to decrease with increasing amounts of WO_3 . The relative intensities of the typical lines of (001), (011), and (111) are shown by the solid lines in Fig. 3, the intensities of which depend on the La^{3+} ion concentration in this type of perovskite structure. It is impossible, however, to neglect the effect of high scattering power of the W^{6+} ion on intensity. In this respect the partial substitution of Mo^{6+} ions is more favored since the two scattering powers of the Nb^{5+} and Mo^{6+} ions are nearly equal. Unfortunately attempts to prepare solid solutions in the $\text{La}_{1/3}\text{NbO}_3\text{--MoO}_3$ system failed. In order to examine the cation distribution, the theoretical intensities of superstructure lines were calculated using Iyer's crystal data,²⁾ the results of which are shown by the broken lines in Fig. 3 in comparison with the observed values. Good agreement between the calculated and the observed intensities was obtained. Therefore, the ordered ar-

Fig. 1. Crystal structure of $\text{La}_{1/3}\text{NbO}_3$.Fig. 2. Lattice constants in the $\text{La}_{1/3}\text{NbO}_3\text{--WO}_3$ system.

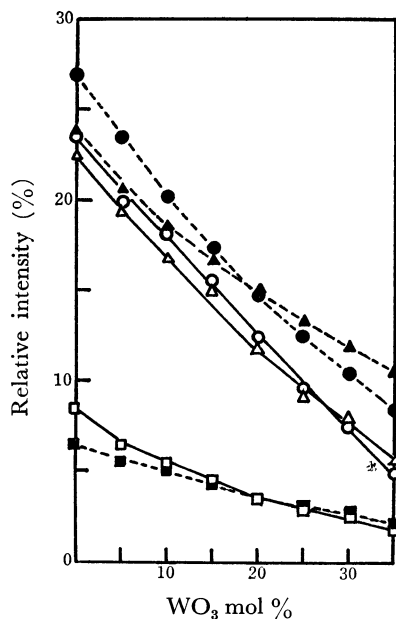


Fig. 3. Observed intensities *vs.* those calculated for typical superstructure lines in this system.

○: $I_{\text{obsd}}(001)$, ●: $I_{\text{calcd}}(001)$,
 △: $I_{\text{obsd}}(011)$, ▲: $I_{\text{calcd}}(011)$,
 □: $I_{\text{obsd}}(111)$, ■: $I_{\text{calcd}}(111)$.

$I_{\text{obsd}}(001)$ represents observed intensity of (001) line, and $I_{\text{calcd}}(111)$ represents the sum of the calculated intensities of (111) and (003) lines which appeared at the nearly same value of 2θ .

range of La^{3+} ions as shown in Fig. 1 is completely maintained in the solid solution in the $\text{La}_{1/3}\text{NbO}_3$ – WO_3 system. This is quite different from the situations showing simple perovskite phases of composition $\text{La}_{1-x}\text{TiO}_3$ ($x=0.03$ – 0.30), except $\text{La}_{2/3}\text{TiO}_3$ ($x=0.33$) with the ordered structure.

Figure 4 shows the variation of lattice constants of $\text{La}_{1/3}\text{NbO}_3$ with temperature. The orthorhombic subcell changes into a tetragonal one above *ca.* 190 °C. The $c/2a$ ratio is almost independent of temperature. For the 20 mol % WO_3 sample, the crystal lattice gradually expands with temperature, and the tetragonality remains almost unchanged. The superstructure lines for both samples did not disappear even at 1000 °C. Generally, the low symmetry of ordinary perovskites is expected to improve at high temperatures. It is likely that $\text{La}_{1/3}\text{NbO}_3$ and its solid solutions differ from ordinary perovskites in the nature of tetragonal distortion. In $\text{La}_{1/3}\text{NbO}_3$ two Nb^{5+} ions are located at $\pm(1/2, 1/2, z)$, $z=0.262$. The intensities of the superstructure lines are relatively sensitive to the atomic

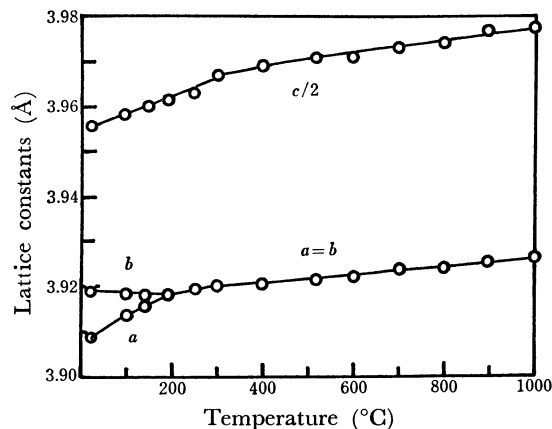


Fig. 4. Temperature dependence of lattice constants for $\text{La}_{1/3}\text{NbO}_3$.

z parameter from the structure factor. With increase in the z value, the (001) line tends to become weaker while the (011) line becomes stronger. For example, the calculated values of (001) and (011) lines for the 20 mol % WO_3 sample were 16.0 and 9.9 at $z=0.25$, and 14.7 and 14.9 at $z=0.262$, respectively. The corresponding observed values were 12.4 and 11.8, respectively. It becomes apparent that the differences between the calculated and observed intensities decrease by the deviation of Nb^{5+} and W^{6+} ions from the B positions of $\pm(1/2, 1/2, 1/4)$. In terms of charge balance, this antiparallel displacement of Nb^{5+} and W^{6+} ions with high valency is considered an important role for the formation of this type of superstructure, where the corner sites are occupied by corresponding amounts of La^{3+} ion and the vertical edge centre sites are vacant. It might be related to the origin of superstructure that the tetragonality in $\text{La}_{1/3}\text{NbO}_3$ and its solid solution remains nearly unchanged with temperature.

References

- 1) H. P. Rooksby, E. A. D. White, and S. A. Langston, *J. Am. Ceram. Soc.*, **48**, 447 (1965).
- 2) P. N. Iyer and A. J. Smith, *Acta Crystallogr.*, **23**, 740 (1967).
- 3) M. Kestigian and R. Ward, *J. Am. Chem. Soc.*, **77**, 6199 (1955).
- 4) D. Ridgley and R. Ward, *J. Am. Chem. Soc.*, **77**, 6132 (1955).
- 5) M. Abe and K. Uchino, *Mater. Res. Bull.*, **9**, 147 (1974).
- 6) Y. Torii and K. Hasegawa, *Bull. Chem. Soc. Jpn.*, **50**, 2638 (1977).